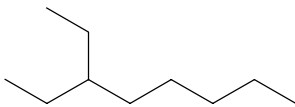
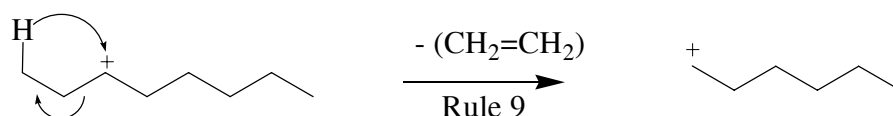
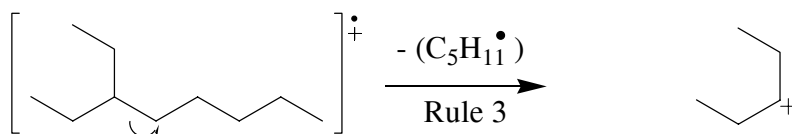
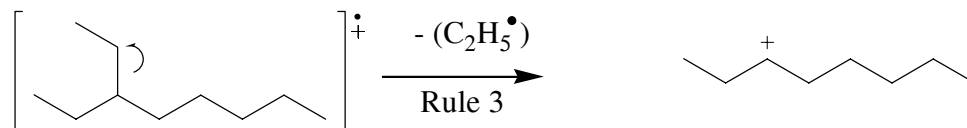
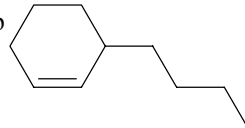


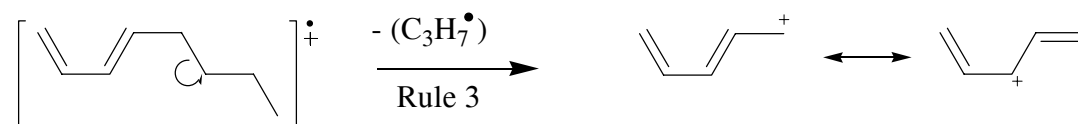
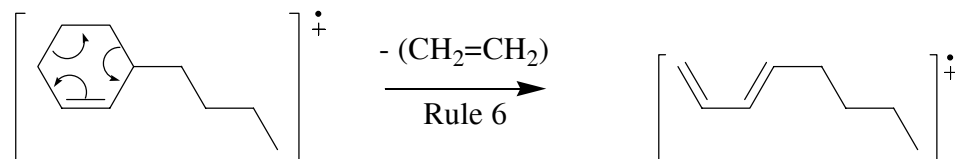
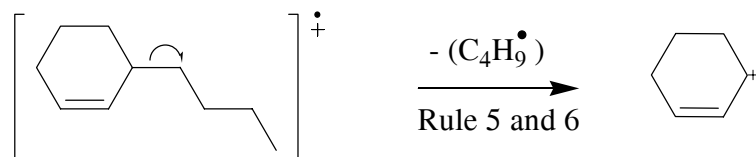
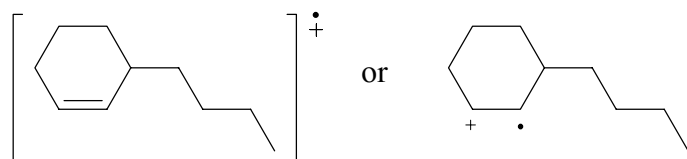
Answers for Student Exercises 1.1 to 1.5

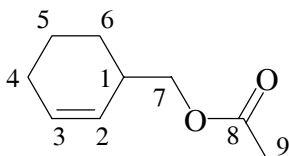
1

<div style="text-align: center;">  <p>a</p> <p>3-ethyloctane</p> <p><math>C_{10}H_{22}</math></p> <p><math>C_nH_mX_xN_yO_z</math></p> <p>Index = <math>(n) - (m/2) - (x/2) + (y/2) + 1 =</math></p> </div>	Atom Type	# of Atoms	Mass	Total Mass
	Hydrogen	22	1.00783	22.17226
	Carbon	10	12.00000	120.00000
	Nitrogen	0	14.0031	0.00000
	Oxygen	0	15.9949	0.00000
	Sulfur	0	31.9721	0.00000
	Bromine	0	78.9183	0.00000
	IHD			
	Exact Mass			142.17226



<div style="text-align: center;">  <p>b</p> <p>3-butylcyclohex-1-ene</p> <p><math>C_{10}H_{18}</math></p> <p><math>C_nH_mX_xN_yO_z</math></p> <p>Index = <math>(n) - (m/2) - (x/2) + (y/2) + 1 =</math></p> </div>	Atom Type	# of Atoms	Mass	Total Mass
	Hydrogen	18	1.00783	18.14094
	Carbon	10	12.00000	120.00000
	Nitrogen	0	14.0031	0.00000
	Oxygen	0	15.9949	0.00000
	Sulfur	0	31.9721	0.00000
	Bromine	0	78.9183	0.00000
	IHD			2
	Exact Mass			138.14094





(b) (cyclohex-2-en-1-yl)methyl acetate

3.1 (b) 1,2,3,4,5,6,7 are one spin system, and 9 is the second. 4,4' ; 5,5' ; 6,6' and 7,7' are diastereotopic.

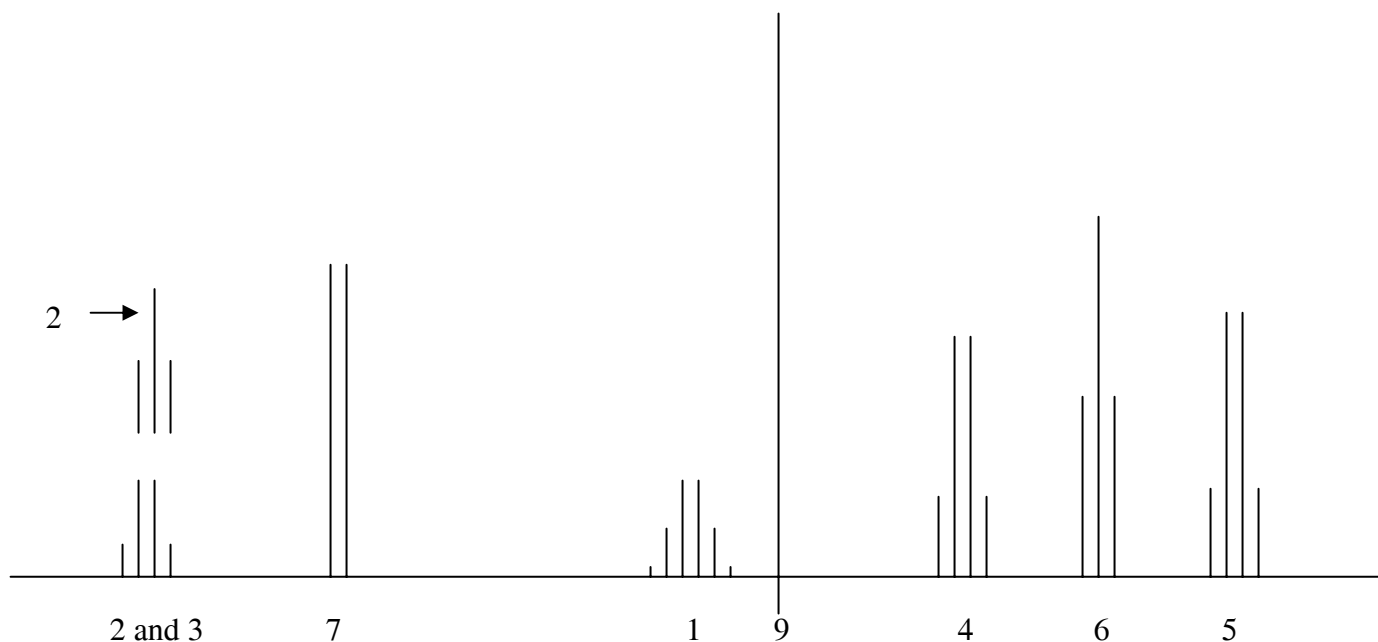
3.2 (b)

Spin	Appr. $\delta$	Table or Calculation	Coupled with	Multiplet (n+1)	Integration
1	$\delta$ 2.80	Appendix A, Chart A.2	6,2,7	Sextet*	1
2	$\delta$ 5.59	Appendix D, Chart D2	3,1	Triplet*	1
3	$\delta$ 5.59	Appendix D, Chart D2		Quartet*	1
4	$\delta$ 1.96	Appendix D, Chart D2		Quartet**	2
5	$\delta$ 1.65	Appendix D, Chart D2	4,6	Quartet**	2
6	$\delta$ 1.80	Appendix A, Chart A.2	1,5	Triplet**	2
7	$\delta$ 4.05	Appendix A, Chart A.2	1	Doublet**	2
9	$\delta$ 2.10	Appendix A, Chart A.1	none	Singlet	3

\* Coupling constants may not be the same, but make the assumption that they are.

\*\*These are diastereotopic, and are coupled to diastereotopic protons, they would not be first order. We will assume they are first order for the drawn spectrum.

3.3 (b)



Answers for Student Exercise 3.10B

Spin System  $ab_2c_3$

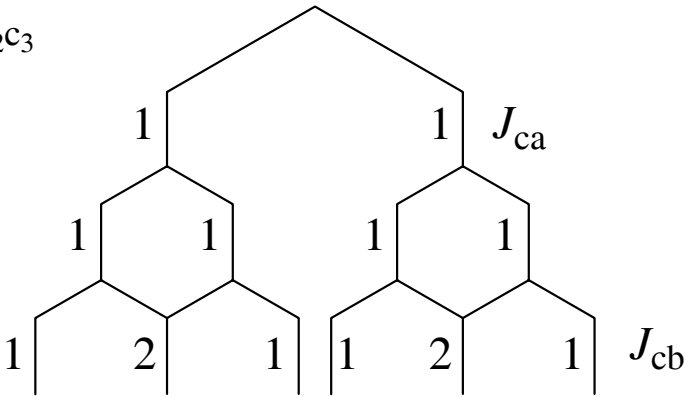
Coupling

a-b 6 Hz

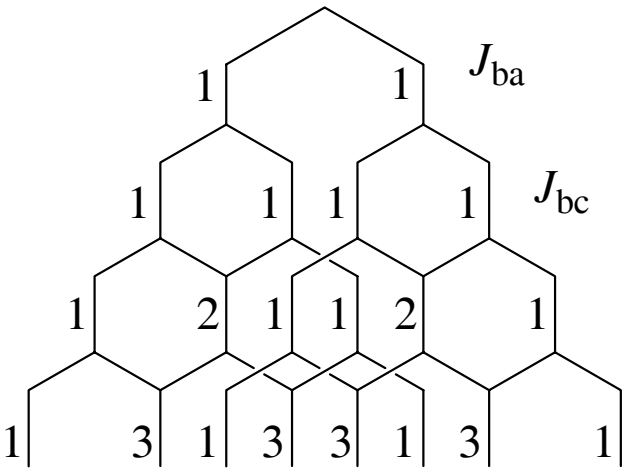
b-c 4Hz

a-c 9 Hz

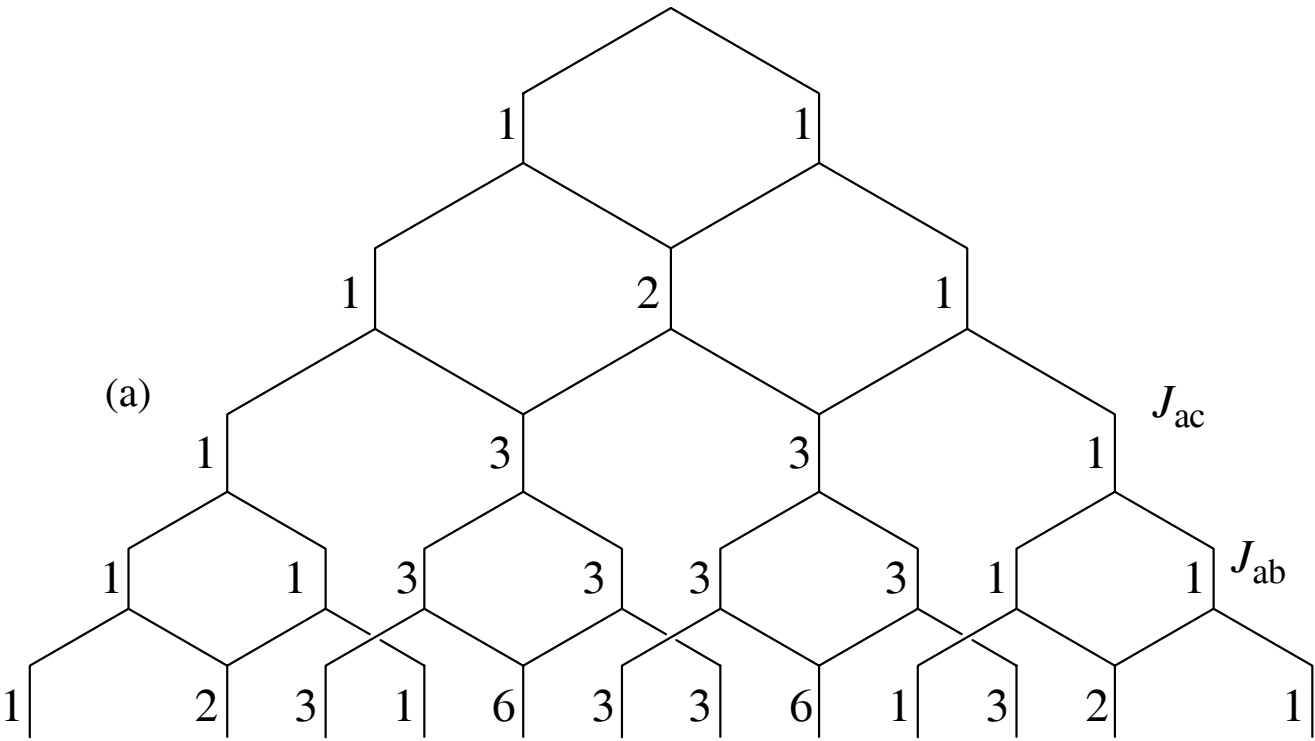
(c)



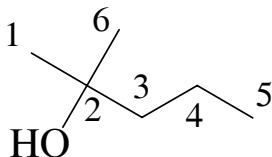
(b)



(a)

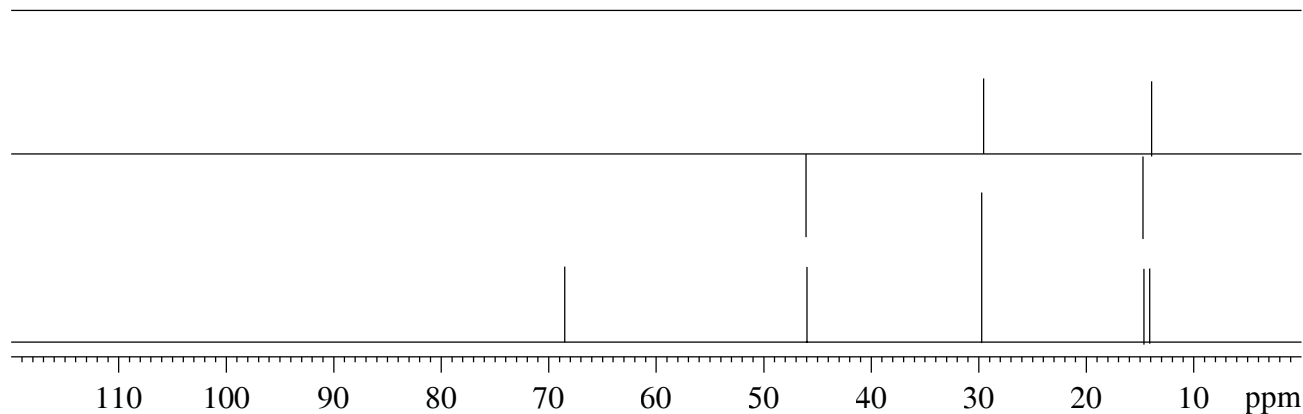


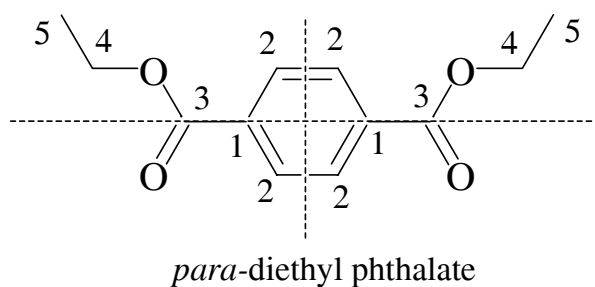
Answers for Student Exercise 4.2-3 (i)



(i) 2-methylpentan-2-ol

Atom	Base	$\alpha$	$\beta$	$\gamma$	$\delta$	$\epsilon$	Other	Calc Shift	Comments
C1	methane	2	6,3,OH	4	5		$1^\circ(4^\circ)$		The base and shifts are from Table 4.4 Increments from table 4.6 See Table 4.14 for comparison
	-2.5	9.1	$2(9.4)+10$	-2.5	0.3		-3.4		
	-2.5	9.1	28.8	-2.5	0.3		-3.4	29.8	
C2	methane	1,6,3,OH	4	5			$4^\circ(2(1^\circ)+2^\circ)$		
	-2.5	$3(9.1)+48$	9.4	-2.5			$(2)(-1.5)+-8.4$		
	-2.5	75.3	9.4	-2.5			-11.4	68.3	
C3	methane	2,4	1,6,OH,5				$2^\circ(4^\circ)$		
	-2.5	$9.1(2)$	$3(9.4)+10$				-7.2		
	-2.5	18.2	38.2				-7.2	46.7	
C4	methane	3,5	2	1,6,OH					
	-2.5	$2(9.1)$	9.4	$2(-2.5)+(-5)$					
	-2.5	18.2	9.4	-10				15.1	
C5	methane	4	3	2	1,6				
	-2.5	9.1	9.4	-2.5	$(2).3$				
	-2.5	9.1	9.4	-2.5	0.6			14.1	
C6	methane	2	6,3,OH	4	5		$1^\circ(4^\circ)$		
	-2.5	9.1	$2(9.4)+10$	-2.5	0.3		-3.4		
	-2.5	9.1	28.8	-2.5	0.3		-3.4	29.8	





There are two mirror images. There are 5 nonequivalent carbons, and there are 3 distinct proton signals, 2,4,5 with a integration ratio of 4:4:6, or 1:1:3.

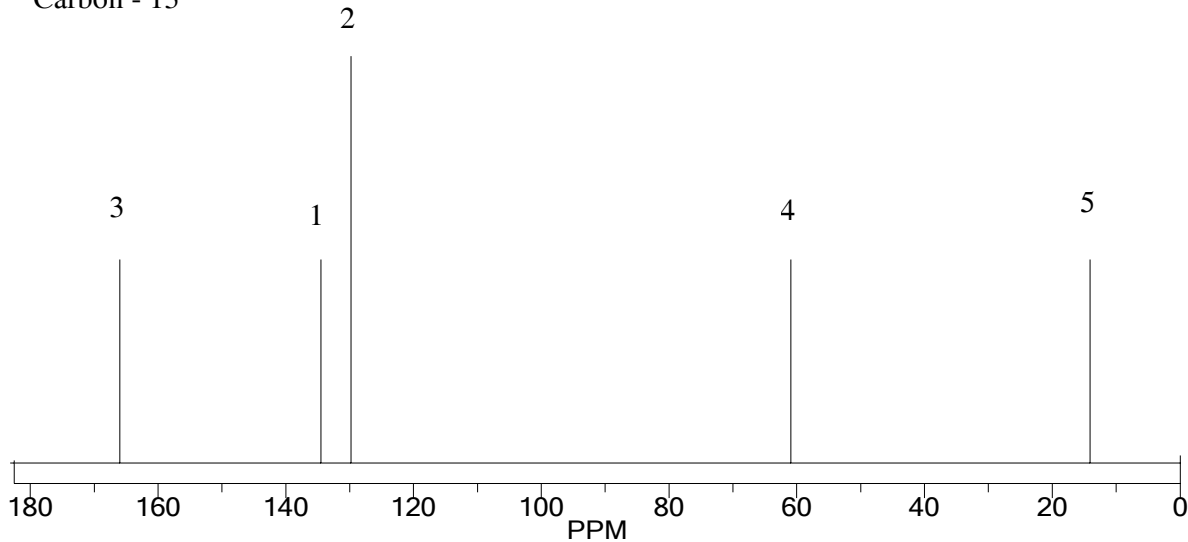
DEPT 90

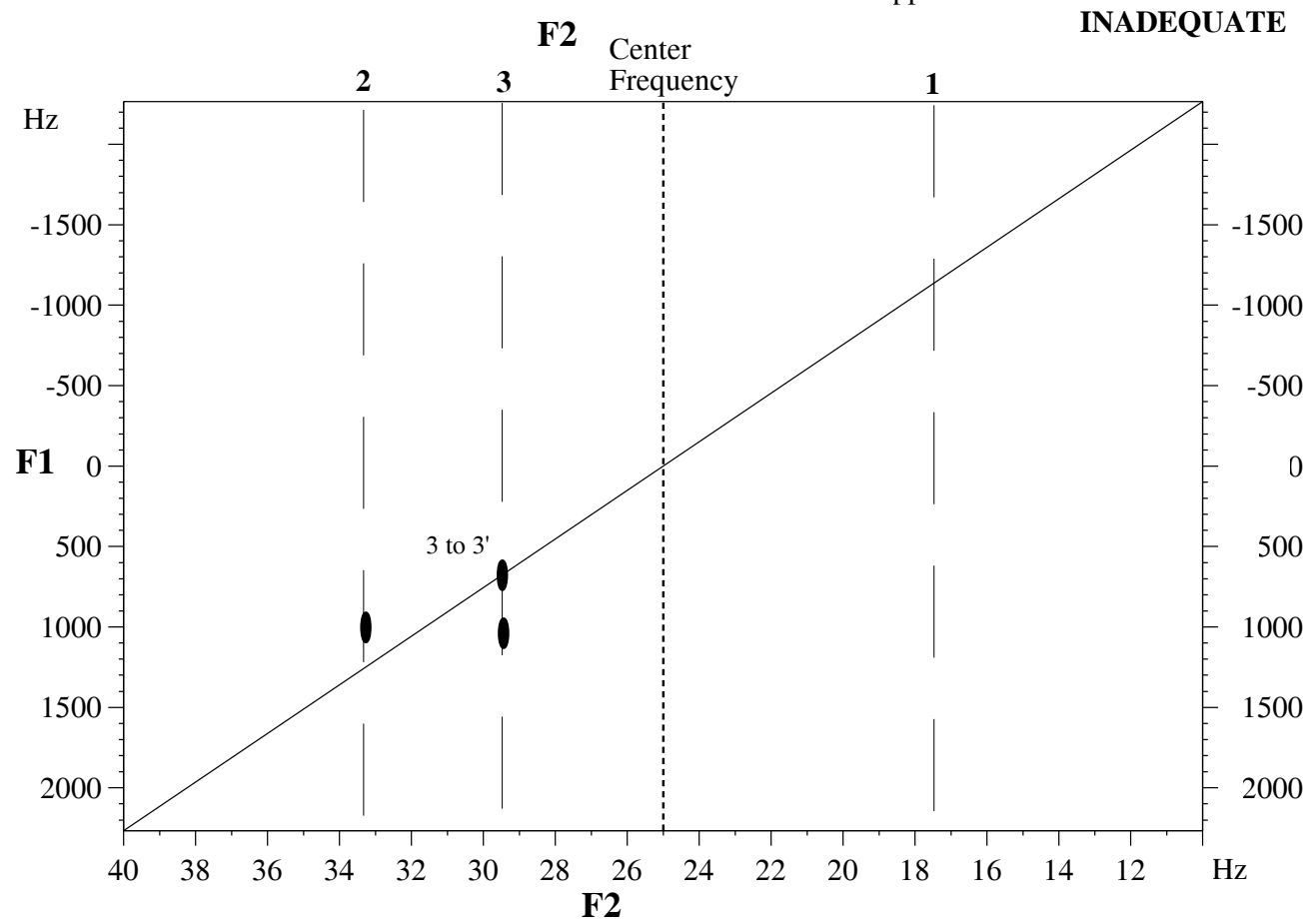
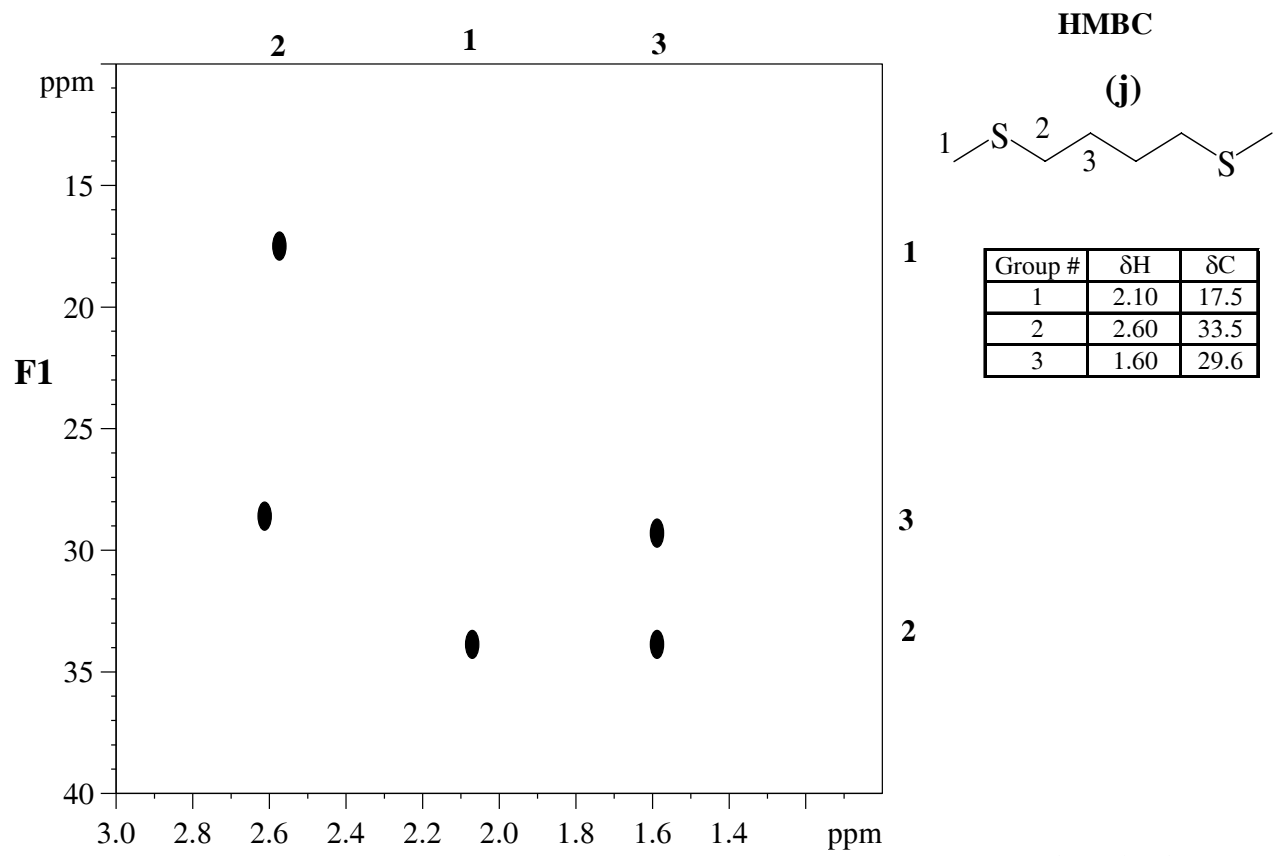


DEPT 135

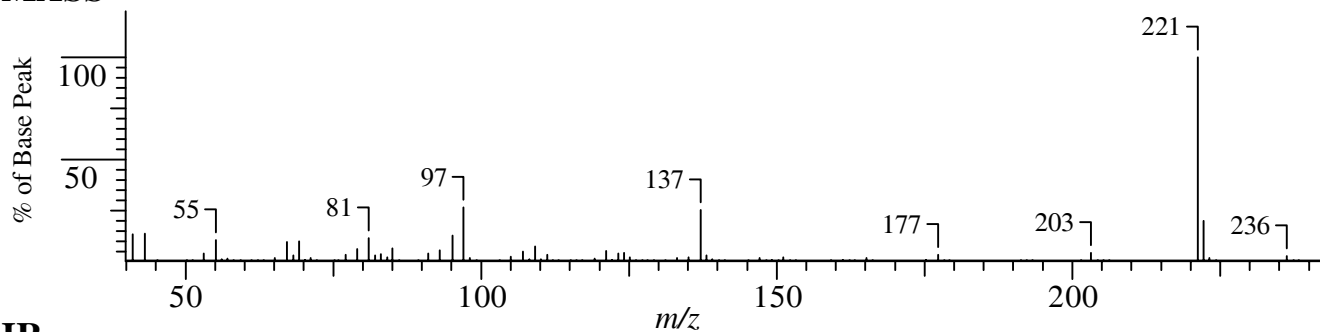


Carbon - 13

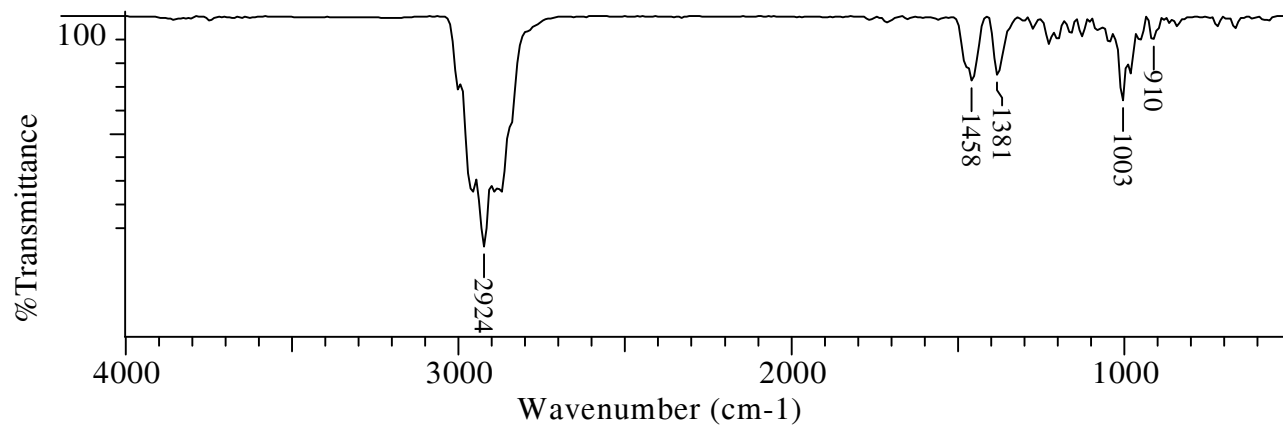




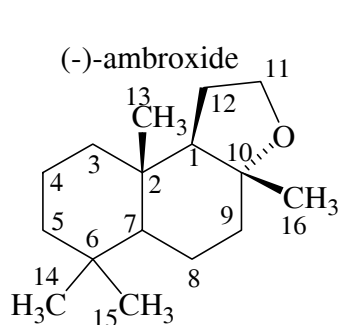
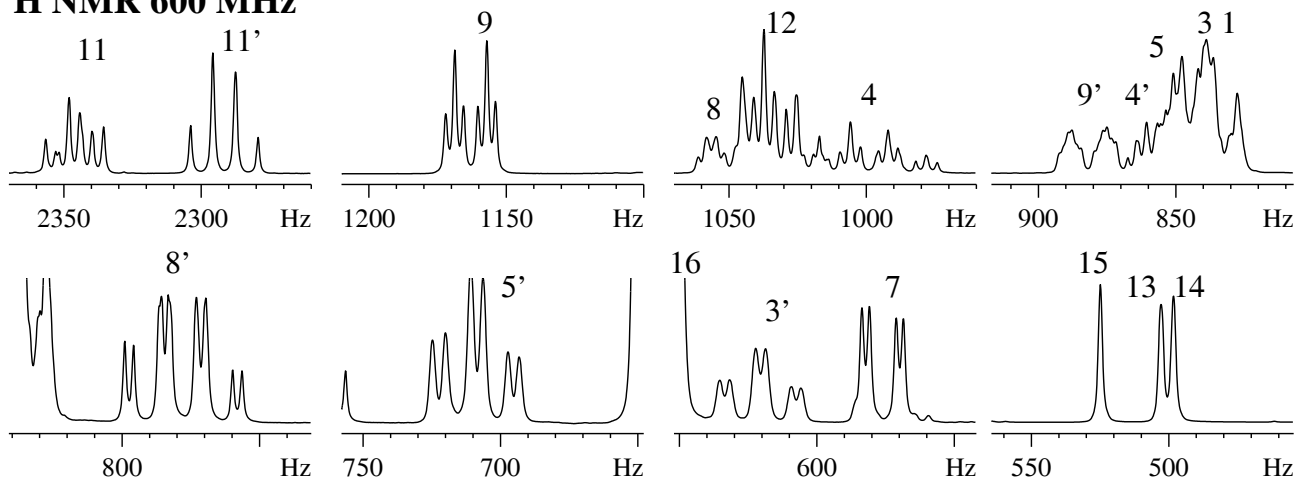
# MASS



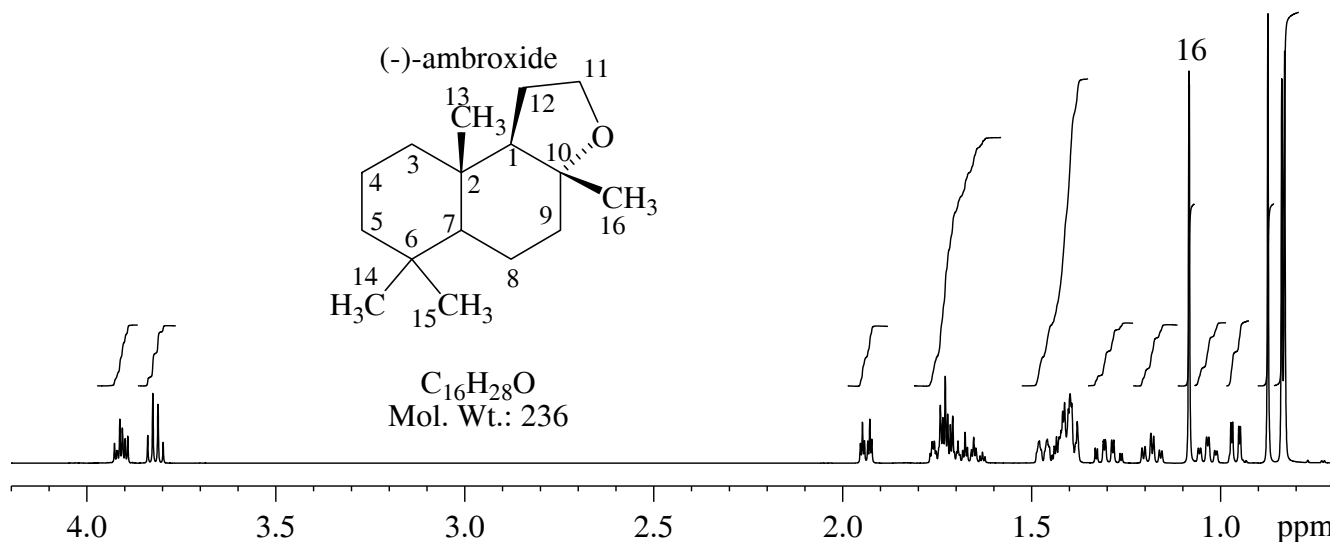
# IR



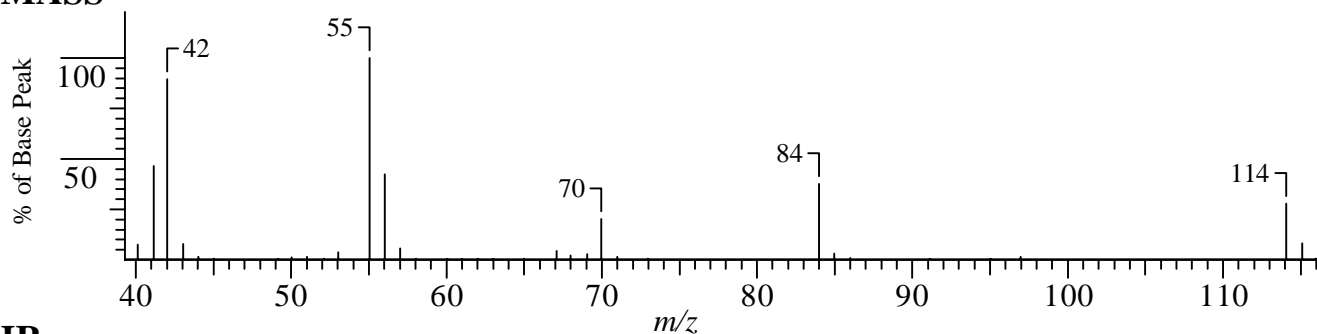
# $^1\text{H}$ NMR 600 MHz



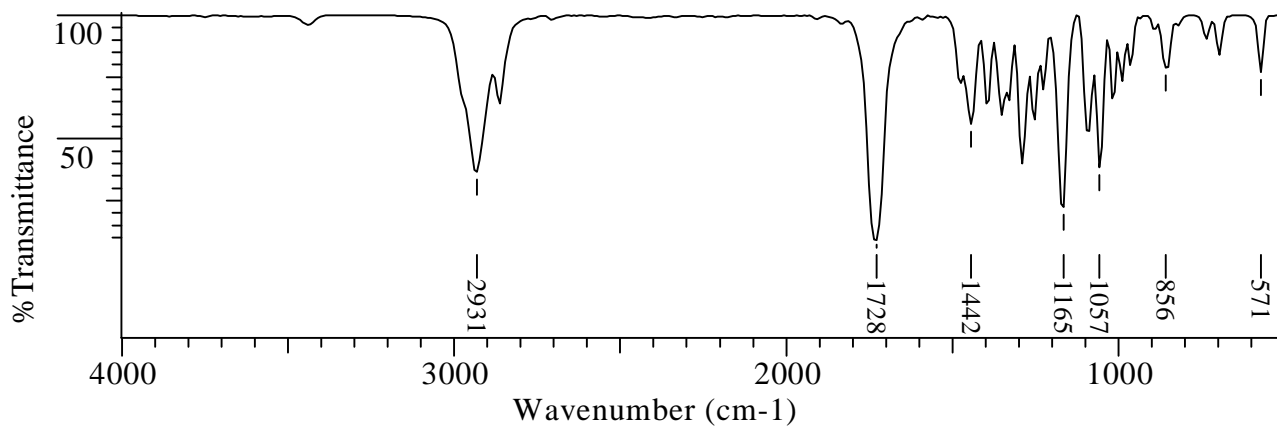
$\text{C}_{16}\text{H}_{28}\text{O}$   
Mol. Wt.: 236



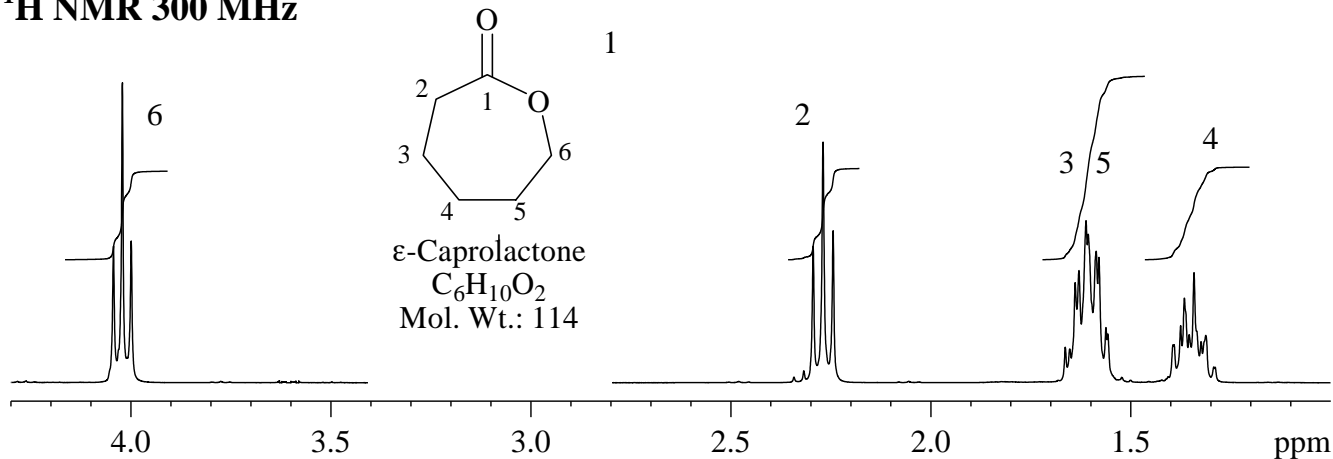
**MASS**



**IR**



**$^1\text{H}$  NMR 300 MHz**



**$^{13}\text{C}/\text{DEPT}$  NMR 75.5 MHz**

